α-HUMULENE DERIVATIVES INCLUDING A SESQUITERPENE ACID WITH A REARRANGED CARBON SKELETON FROM LYCHOPHORA COLUMNARIS*

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Abstract—An investigation of Lychnophora columnaris afforded in addition to known compounds six new α -humulene derivatives and a rearranged sesquiterpene with a new carbon skeleton. These compounds and several sesquiterpene lactones isolated from this species seem to be of chemotaxonomic importance. The structures were elucidated by spectroscopic methods and a few chemical transformations including partial synthesis.

INTRODUCTION

So far eight species of the genus Lychnophora have been investigated chemically. As in other Brazilian genera, namely Eremanthus, Piptolepis, Proteopsis and Vanillosmopsis, all belonging to the subtribe Lychnophorinae [1], most species contain furanoheliangolides [2-11], while eremanthin has also been isolated from six Lychnophora sp. [6, 7, 11] which may be characteristic for the subtribe. Oxidized carvophyllene derivatives are also typical for Lychnophora [6, 11-13], while other constituents have been isolated from only one or two species. We have now investigated Lychnophora columnaris, which again afforded a furanoheliangolide, eremanthin and related lactones, while the caryophyllene derivatives were replaced by oxidized α -humulene derivatives, six of them not previously isolated. Furthermore, a sesquiterpenic acid with a new carbon skeleton was isolated.

RESULTS AND DISCUSSION

The aerial parts of Lychnophora columnaris Mattf. afforded lupeol and its acetate, stigmasterol, eremanthin (21) [14], lychnopholide (24) [6] and α -humulene (1) together with several compounds derived from the latter, the alcohol 2 [15], the corresponding acetate 4 and aldehyde 5 as well as the acid 3. Furthermore, the epoxides 6-9 were present, only 8 having been previously isolated [16]. The structure of 3 followed from the ¹H NMR spectrum (Table 1), which showed signals of an olefinic methyl and of two tertiary methyls. The IR spectrum indicated the presence of an unsaturated acid, which was supported by a

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downfield olefinic broadened triplet at δ 6.09 in the ¹H NMR spectrum. A second broadened triplet at δ 4.98 was coupled with the olefinic methyl and with a broadened doublet at 1.93. Irradiation of a broadened double triplet at 2.69 collapsed the triplets at 6.09 and 2.22 to singlets, while irradiation of the olefinic signal at 5.60 allowed the assignment of the signals at 5.17 and 2.89. The resulting sequences led to the proposed structure 3, which was confirmed by reduction, which afforded 2, identical with the natural alcohol. The ¹³C NMR spectrum also was in good agreement with the structure (see Experimental). The ¹H NMR spectral data of 4 (Table 1) were close to those of 2, only the H-14 signals being shifted drastically downfield. Compound 5 was prepared previously by oxidation of 2 [15]. The spectral data of both aldehydes were identical. The ¹H NMR spectral data of 6 (Table 1) indicated an epoxide of 2. All signals could be assigned by spin decoupling. However, due to the flexibility of the 11-membered ring system the stereochemistry of the epoxide could not be determined with certainty. We therefore have epoxidized 2, which afforded two mono- and two diepoxides. The main product was identical with 6. As inspection of models showed, the epoxidation of the 1, 10-double bond should be achieved from one side only leading to 6. Obviously the preferred reaction of this double bond was influenced by neighbouring group participation of the hydroxyl. The ¹H NMR spectral data of the second monoepoxide (Table 1) led to the structure 18. The stereochemistry at C-4 and C-5 was assigned indirectly by consideration of the data of the two diepoxides. The ¹H NMR spectral data of these compounds (19 and 20) (Table 1) clearly indicated that they differed in the stereochemistry at C-4 and C-5 only. The downfield shift of the H-9 α signal in the spectrum of 20 could be explained by a deshielding effect of the 1, 10-epoxide. However, this required

Scheme 1.

^{*}Numbering as in germacranes.

Table 1. ¹ H NMI	R spectral data	a of compou	nds 3, 4	, 6, 7 and	18-20 (400 MHz,	CDCl ₃ ,	TMS as	internal
			stan	dard)				

	3	4	6	7	18	19	20
H-1	6.09 <i>br t</i>	5.44 <i>br</i> t	2.69dd	2.92dd	5.36 <i>br dd</i>	2.79dd	2.87 <i>dd</i>
Η-2α]		2.17 <i>dddd</i>	2.07 <i>m</i>	2.28m	2.02ddd	2.21 <i>dddd</i>
}	2.69brdt	2.17 <i>br dt</i>					4 40 1 1 1 1
H-2β]	J		1.50m	1.55 <i>m</i>	2.22 <i>m</i>	1.67 <i>dddd</i>	1.48 <i>dddd</i>
$H-3\alpha$	j		2.03 <i>m</i>		1.41 <i>m</i>	1.25 <i>ddd</i>	0.98 <i>ddd</i>
}	2.22brt	2.11 <i>br t</i>		2.27 <i>m</i>			
Η-3β	J		2.25 <i>br ddd</i> J		2.13 <i>ddd</i>	2.23ddd	2.15 <i>ddd</i>
H-5	4.98 <i>br t</i>	5.12 <i>br t</i>	4.99 <i>br dd</i>	4.99 <i>br dd</i>	2.57 <i>br d</i>	2.67 <i>d</i>	2.46 <i>d</i>
Η-6α)		1.87 <i>dd</i>		1.62 <i>br d</i>	1.73 <i>d</i>	1.61 <i>d</i>
}	1.93 <i>br d</i> }	1.91 <i>br d</i>		1.90 <i>br d</i>			
Η-6β	J		1.98 <i>dd</i>	·	1.41 <i>br dd</i>	1.43 <i>dd</i>	1.37 <i>dd</i>
H-7	5.17 <i>d</i>	5.37d	5.19d	5.23 <i>d</i>	5.40 <i>d</i>	5,54m	5.34d
H-8	5.60 <i>dt</i>	5.78dt	5.31 <i>ddd</i>	5.41 <i>ddd</i>	5.78ddd	5.5 4m	5.54 <i>ddd</i>
Η-9α	J		2.98dd	3.05 <i>dd</i>	2.75dd	2.88 <i>br d</i>	3.06 <i>dd</i>
}	2.89brd	2.61 <i>br d</i>	2.7000	3.00	2		•
н-9β	J		1.50 <i>m</i>	1.55 <i>m</i>	2.69 <i>dd</i>	1.79 <i>dd</i>	1.53dd
H-12]		1.10 <i>s</i>	1.10 <i>s</i>	1.18s	1.16 <i>s</i>	1.19 <i>s</i>
}	1. 06 s	1.03 <i>s</i>		4.05		1.00	1.00
H-13 J	J		1.08 <i>s</i>	1.07 <i>s</i>	1.03 <i>s</i>	1.08s	1.08 <i>s</i>
H-14)		3.91 <i>dd</i>		4.22 <i>d</i>	3.89d	3.87 <i>d</i>
	}	4.76 <i>s</i>	}	9.35 <i>d</i>			
H-14	J		3.57d	1.461	4.12 <i>d</i>	3.72d	3.58dd
H-15	1.45 <i>d</i>	1.43 <i>d</i>	1.54 <i>br s</i>	1.46 <i>br s</i>	1.19 <i>s</i>	1.19 <i>s</i>	1.30s
ОН		_	1.72 <i>br</i>	_	_	1.25 <i>br s</i>	1.61 <i>dd</i>

J(Hz): Compound 3: 1,2=8; 2,3=6.5; 5,6=7; 5,15=1; 7,8=16; 8,9=7.5; compound 4: 1,2=5,6=8,9=7.5; 2,3=7; 5,15=1; compound 6: $1,2\alpha=4$; $1,2\beta=10$; $2\alpha,2\beta=14$; $2\alpha,3\alpha=4$; $3\alpha3\beta=12.5$; $5,6\alpha=6$; $5,6\beta=8$; $6\alpha,6\beta=14$; 7,8=16; $8,9\alpha=5$; $8,9\beta=10$; $9\alpha,9\beta=13$; 14,14'=11.5; 14,OH=6; compound 7: $1,2\alpha=3.5$; $1,2\beta=10$; 5,6=7; 7,8=16; $8,9\alpha=5$; $8,9\beta=10$; $9\alpha,9\beta=13$; $9\alpha,14=1$; compound 18: $1,2\sim7$; $2\alpha,3\beta=4.5$; $2\beta,3\beta=2.5$; $3\alpha,3\beta=13$; $5,6\beta=10$; $6\alpha,6\beta=14$; 7,8=16; 8,9=7; $9\alpha,9\beta=14$; 14,14'=12; compound 19: $1,2\alpha=1$; $1,2\beta=10$; $2\alpha,2\beta=15$; $2\alpha,3=4$; $2\beta,3\alpha=12$; $2\beta,3\beta=4$; $3\alpha,3\beta=14$; $5,6\beta=10$; $6\alpha,6\beta=9\alpha,9\beta=14$; 14,14'=12; compound 20: $1,2\alpha=5$; $1,2\beta=10$; $2\alpha,2\beta=14$; $2\alpha,3\alpha=13$; $2\alpha,3\beta=5$; $2\beta,3\alpha=5$; $2\beta,3\beta=2$; $3\alpha,3\beta=13$; $5,6\beta=10$; $6\alpha,6\beta=14$; 7,8=15.5; $8,9\alpha=5$; $8,9\beta=10$; $9\alpha,9\beta=13$; 14,14'=12; 14,OH=8; 14',OH=4.

the proposed configuration with a changed conformation. The presence of a hydrogen bridge was supported by the clear couplings of the hydroxyl signal. Furthermore, the presence of a rigid conformation of 20 was supported by the unusually high melting point and low solubility. As expected the 7, 8-double bond was not epoxidized, while the isolation of isomeric 4, 5-epoxides indicated flexible conformations of 2, which allowed a reaction of the 4, 5-double bond from both sides. The shift differences of the H-9 α signals in the spectra of 19 and 20 can be used for the assignment of the stereochemistry of 18. As the signal of H-9 α in the ¹H NMR spectrum of 18 was not shifted downfield, the proposed configuration was very likely. The acid 9 could be obtained pure only as its methyl ester. The ¹H NMR spectral data (Table 2) showed that the 7, 8-double bond of 3 was epoxidized, as the downfield signals of the olefinic protons were replaced by a narrowly splitted doublet at δ 2.33 and a threefold doublet at 3.01. Though the other signals, which could be assigned by spin decoupling, were slightly different, their couplings only agreed with the proposed structure. Though the stereochemistry at C-7 and C-8 could not be established with certainty, the proposed one was most likely, if models were inspected. We have named compound 9 lychnocolumnic acid.

Finally, a further acid with an additional acetoxyl group was present in the plant extract. The molecular formula and the 'H NMR spectral data (Table 2) indicated the presence of a monocyclic triene. However, the 'H NMR signals clearly showed the presence of five methyl groups, one carboxyl, one olefinic, one acetoxy methylene and two tertiary methyl groups, a rearranged humulene derivative had to be assumed. Careful spin decoupling led to the structure 14, which was supported by the 'H NMR spectral data of the corresponding methyl ester 15 and those of 16 and 17 obtained by acid catalysed hydrolysis and methanolysis respectively. These reactions further established the allylic nature of the acetoxyl group. The stereochemistry of the 4, 5-

Table 2. ¹ H NMR spectral data of compounds 10 and 14-17 (400 MHz, CDCl ₃ , TMS	S as
internal standard)	

	10	14	Δ^*	15	16	17
H-1	5.86dd	6.04 <i>br dd</i>	0.01	5.87 <i>br dd</i>	5.89 <i>br dd</i>	5.86 <i>br dd</i>
$H-2\alpha$	2.52dddd	2.49br d	0.21	2.42 <i>br d</i>	2.40brd	2.43brd
$H-2\beta$	2.82 <i>dddd</i>	2.92m	0.23	2.82m	2.75m	2.85m
Η-3α	2.15ddd	2.32 <i>br dd</i>	0.1	2.31 <i>br dd</i>	2.33 <i>br dd</i>	2.30br dd
$H-3\beta$	2.88brd	2.24 <i>ddd</i>	0.1	2.20ddd	2.27ddd	2.23 <i>ddd</i>
H-5	4.98 <i>br d</i>					
Η-6α	2.28brd			_		
Η-6β	2.20dd		_	_		
H-7	2.33 <i>d</i>	5.17 <i>d</i>	0.17	5.18d	5.15d	5.16d
H-8	3.01 <i>ddd</i>	5.77 dt	0.16	5.77 dt	5.70dt	5.71 <i>dt</i>
Η-9α	3.08dd	2.85 <i>br dd</i>	0.23	2.81 <i>br dd</i>	2.69 <i>br dd</i>	2.80br da
Η-9β	1.72dd	2.97 <i>br dd</i>	0.13	2.96 <i>br dd</i>	2.86 <i>br dd</i>	2.93 <i>br dd</i>
H-11		5.02 <i>br d</i>	0.11	5.02 <i>br d</i>	4.95br d	5.08 <i>br d</i>
H-11'		5.16d	0.11	5.16d	3.45d	3.98d
H-12	1.08s	1.10s	0.07	1.10s	1.11 <i>s</i>	1.13 <i>s</i>
H-13	0.73s	1.02 <i>s</i>	0.06	1.02s	1.01 <i>s</i>	1.04s
H-15	1.53 <i>br s</i>	1.67 <i>br s</i>	0.14	1.63 <i>br s</i>	1.53 <i>br s</i>	1.54brs
OMe	3.76s			3.76s	3.75 <i>s</i>	3.76s
					3.11 <i>s</i>	_
OAc	_	2.03 <i>s</i>	0.03	2.03s		

^{*}Δ-values after addition of Eu(fod)₁.

J(Hz): Compound 10: 1,2α = 8.5; 1,2β = 8.5; 2α,2β = 13; 2α,3α = 5; 2α,3β = 12; 2β,3α = 2.5; 2β,3β = 4; 3α,3β = 12.5; 5,6α = 2; 5,6β = 10.5; 6α,6β = 15; 7,8 = 2.5; 8,9α = 4; 8,9β = 9; 9α,9β = 13; compounds 14–17: 1,2α = 5; 1,2β = 10; 2α,2β = 13; 2α,3α = 5; 2β,3α = 5; 2β,3β = 12; 3α,3β = 12.5; 7,8 = 16; 8,9 = 7; 9α,9β = 15; 11,11′ = 11

double bond could not be determined with certainty. However, inspection of models showed that the proposed orientation was most likely. The chemical shifts of H-3, H-12 and H-13 and the mild reaction conditions of the methanolysis agreed with this assumption. Inspection of models led to the proposed assignment of the signals, which were in agreement with the couplings observed, if a conformation was assumed with the C-10 methyl in plane and the C-4 above the plane. This would explain the differences in the observed chemical shifts of H-9 α and H-9 β as well as of H-12 and H-13. The structure was further supported by the ¹³C NMR signals. Biogenetically 14 may be formed starting with 3 by allylic oxidation leading to 11, which after epoxidation could be rearranged to 13. Elimination of H₂O, reduction and acetylation could then give 14 (Scheme 1). Compound 14 seems to be derived from a so far unknown carbon skeleton, for which we propose the name lychnene. Therefore compound 14 was named 11 - acetoxylychnen - 14 - oic acid.

The roots afforded tridecapentaynene, trideca - 3,5,7,9 - tetrayn - 1,11 - diene, lupeol and its acetate, 21, 22 [11], 23 [17] and 24.

The chemistry of this species again showed that furanoheliangolides and erementhin may be characteristic for the subtribe *Lychnophorinae*. The α -humulene derivatives seem to replace the caryophyllene derivatives isolated from several other species.

As caryophyllene and α -humulene are biogenetically closely related, this change is a very small one. Further investigations may show if enzymes for the oxidation of simple sesquiterpenes are widespread in this genus. So far derivatives of α -humulene are very rare in the Compositae.

EXPERIMENTAL

The air dried plant material (voucher RMK 8610, deposited in the United States National Herbarium, Washington) was extracted with Et₂O-petrol (1:2) and the resulting extracts were separated by CC (Si gel) and further by repeated TLC (Si gel). Known compounds were identified by comparing the 'H NMR spectra with those of authentic material. The aerial parts (750 g) afforded 100 mg lupeol, 20 mg stigmasterol, 100 mg lupeyl acetate, 100 mg 1, 500 mg 2, 200 mg 3 (Et₂O-petrol, 1:3), 5 mg 4 (Et₂O-petrol, 1:0), 5 mg 5 (Et₂O-petrol, 1:10), 10 mg 6 (Et₂O-petrol, 1:1), 5 mg 7 (Et_2O -petrol, 1:10), 10 mg 8, 10 mg 9 (Et_2O -petrol, 1:1), 20 mg 14 (Et₂O-petrol, 1:1), 20 mg 21 and 100 mg 24. Compounds 9 and 14 were transformed to the corresponding methyl esters by addition of CH₂N₂. The roots (150 g) gave 1 mg tridecapentaynene, 0.1 mg trideca - 3,5,7,9 - tetrayn -1,11 - diene, 100 mg lupeol, 500 mg lupeyl acetate, 4 mg 21, 3 mg 22, 3 mg 23 and 10 mg 24.

 α -Humulen-14-oic acid (3). Colourless crystals, mp 93° (petrol); IR $\nu_{\text{max}}^{\text{CCL}}$ cm⁻¹: 3500–2500, 1690, 1640 (C=CCO₂H), 980 (trans CH=CH); MS m/z (rel. int.): 234.162 [M]⁺ (18) (C₁₅H₂₂O₂), 219 (10), 201 (11), 189 (19), 165 (26), 151 (68), 59

(100); 13 C NMR (CDCl₃) (C-1 through C-15): 151.7 d, 25.3 t, 40.1 t, 132.9 s, 125.3 d, 41.9 t, 142.5 d, 127.5 d, 33.9 t, 133.4 s, 37.4 s, 27.0 q, 27.1 q, 174.0 s, 14.9 q (signals assigned by spin decoupling and gated decoupling). Compound 3 (10 mg) on reduction with LiAlH₄ in Et₂O afforded 2, identical with the natural compound.

14-Acetoxy- α -hymulene (4). Colourless oil, IR $\nu_{\rm max}^{\rm CCl_4}$ cm⁻¹: 1745, 1240 (OAc), 980 (trans CH=CH); MS m/z (rel. int.): 218.167 [M – ketene]⁺ (34) (C₁₅H₂₂O), 203 [218 – Me]⁺ (10), 187 [M – CH₂OAc]⁺ (11), 119 (100).

 α -Humulen-14-al (5). Colourless oil, IR $\nu_{\rm max}^{\rm CCl_4}$ cm⁻¹: 2750, 1675 (C=CCHO), 980 (trans CH=CH); MS m/z (rel. int.): 218.167 [M]⁺ (15) (C₁₅H₂₂O), identical with the aldehyde obtained by MnO₂ oxidation of 2 [15].

14 - Hydroxy - 1α , 10β - epoxy - 1, 10 - dihydro - α - humulene (6). Colourless oil, IR $\nu_{\rm max}^{\rm CCL}$ cm⁻¹: 3600 (OH), 1660, 865 (C=CH), 980 (trans CH=CH); MS m/z (rel. int.): 236.178 [M]⁺ (4) (C₁₅H₂₄O₂), 221 (4), 205 (3), 154 (20), 107 (44), 93 (65), 91 (77), 79 (63), 67 (100), 55 (94).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{+13.5} \frac{578}{+14} \frac{546}{+16} \frac{436 \text{ nm}}{+29}$$
 (CHCl₃; $c = 0.68$).

Partial synthesis of 6. Compound 2 (100 mg) in 2 ml CHCl₃ was stirred for 30 min with 100 mg m-chloroperbenoic acid and 0.5 ml satd NaHCO₃ soln. TLC (Et₂O-petrol, 1:1) afforded 10 mg 2, 10 mg 6 (¹H NMR spectrum identical with that of the natural compound), 5 mg 18, 10 mg 19 and 5 mg 20.

Compound 18: Colourless oil, IR $\nu_{\text{max}}^{\text{CCL}}$ cm⁻¹; 3605 (OH), 980 (trans CH=CH); MS m/z (rel. int.): 236.178 [M]⁺ (3) (C₁₅H₂₄O₂), 221 [M - Mc]⁺ (5), 203 [221 - H₂O]⁺ (5), 121 (68), 119 (63), 93 (90), 79 (89), 67 (89), 55 (100).

Compound 19: Colourless gum, IR $\nu_{\text{CCL}}^{\text{CCL}}$ cm⁻¹: 3600 (OH), 1640, 980 (trans CH=CH); MS (CI, iso-butane) m/z (rel. int.): 253 [M+1]⁺ (100), 235 [253 – H₂O]⁺ (82), 217 [235 – H₂O]⁺ (50).

Compound 20: Colourless crystals, mp 154° (Et₂O), IR $\nu_{\text{max}}^{\text{CCL}}$ cm⁻¹: 3580 (OH), 980 (trans CH=CH); MS (CI, isobutane) m/z (rel. int.): 253 [M+1]⁺ (100), 235 [253 – H₂O]⁺ (87), 217 [235 – H₂O]⁺ (51).

 1α , 10β - Epoxy - 1, 10 - dihydro - α - humulen - 14 - al (7). Colourless oil, IR $\nu_{\rm mcl}^{\rm CCL}$ cm⁻¹: 2730, 1720 (CHO), 980 (trans CH=CH); MS m/z (rel. int.): 234.162 [M]⁺ (8) (C₁₅H₂₂O₂), 219 [M - Me]⁺ (8), 205 [M - CHO]⁺ (7), 187 [205 - H₂O]⁺ (6), 67 (100).

$$[\alpha]_{24^\circ}^{\lambda} = \frac{589}{+13} \frac{578}{+14} \frac{546}{+16} \frac{436 \text{ nm}}{+28}$$
 (CHCl₃; $c = 0.24$).

Lychnocolumnic acid (9). Isolated as its methyl ester 10, colourless gum, IR $\nu_{\text{max}}^{\text{CCL}_{4}}$ cm⁻¹: 1715, 1640 (C=CCO₂Me); MS m/z (rel. int.): 264.172 [M]⁺ (1) (C₁₆H₂₄O₃), 232 [M - MeOH]⁺

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{-32} \frac{578}{-33} \frac{546}{-40} \frac{436 \text{ nm}}{-83}$$
 (CHCl₃; $c = 0.51$).

11 - Acetoxylychnen - 14 - oic acid (14). Colourless gum, IR $\nu_{\rm max}^{\rm CCl_4}$ cm⁻¹: 3500–2500, 1680, 1625 (C=CCO₂H), 1730, 1240 (OAc); MS m/z (rel. int.): 292.167 [M]⁺ (4) (C₁₇H₂₄O₄), 232 [M-HOAc]⁺ (90), 217 [232-Me]⁺ (24), 203 (46), 189 (65), 176 (48), 166 (58), 163 (77), 162 (71), 151 (87), 145 (81), 133 (70), 121 (100), 107 (76), 105 (80), 91 (89), 84 (90), 79 (80), 69 (95).

Preparation of 16 and 17. 15 mg of the methyl ester 15 (prepared by addition of CH₂N₂ in Et₂O, 5 min) in 2 ml MeOH and 0.5 ml 2 N H₂SO₄ was heated for 5 min at 70°. Usual work-up and TLC (Et₂O-petrol, 1:1) afforded 3 mg 16 (¹H NMR see Table 2) and 10 mg 17, colourless gum, IR $\nu_{\rm max}^{\rm CCl_4}$ crel. int.): 278.188 [M]⁺ (3) (C₁₇H₂₆O₃), 246 [M – HOMe]⁺ (21), 231 [246 – Me]⁺ (6), 180 (41), 165 (68), 121 (100). ¹³C NMR (CDCl₃) (C-1 through C-15): 148.4 d, 26.9 t, 40.8 t, 133.8 s, 137.9 s, 41.0 s, 141.4 d, 125.7 d, 33.6 t, 133.9 s*, 83.8 t, 24.7 q, 24.7 q, 174.0 s*, 17.7 q, 51.3 q and 56.3 q (OMe).

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^{*}Not certain, very weak signals.